

Thermionic emission

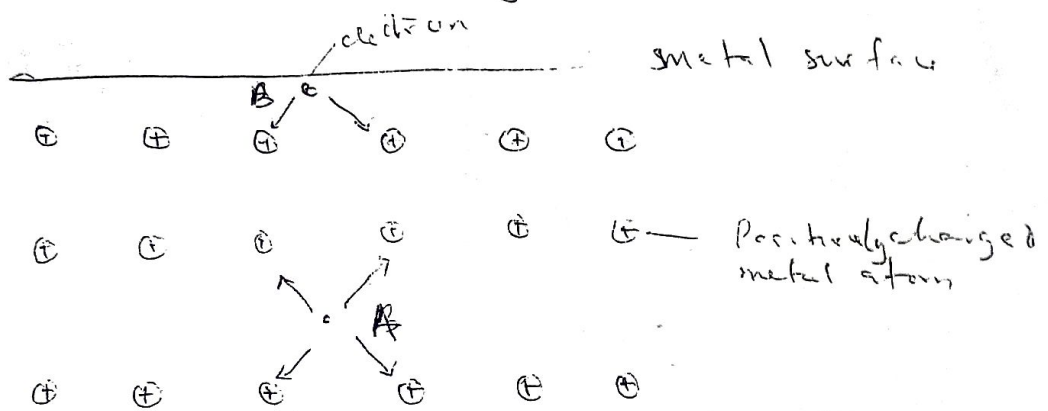
Electron can escape from the ~~surface~~ ^{various} of material in three ways, the chief of these are

- 1) Thermionic emission
- 2) Photoelectric emission
- 3) Secondary emission

Thermionic emission is the ~~most~~ ^{often} ~~often~~ method used to obtain a supply of electrons.

In this process, electrons are made to leave the surface of a metal or a metal coated with certain oxides by the application of heat.

The action can be seen to occur in the following way. The atoms in a metal are regularly spaced forming a rigid lattice structure, see fig 1



Every atom has one or more loosely held electrons which, due to the close packing of atoms in a solid are not attracted to any particular atom but form an 'electron cloud' belonging to the metal as a whole. The atoms, being short of electrons are positively charged and exert attractive forces on the electrons in the inter-atomic spaces.

At points like A inside the metal, the forces on an electron more or less balance; near the surface or B, an electron trying to escape experiences an

inward pull due to the absence of atoms above it. The surface act as a barrier which cannot be penetrated by an electron unless its kinetic energy is increased. If the metal is heated to a high temperature, many of the inter-atomic electrons gets sufficient energy to break out.

The energy required by an electron to enable it to escape from the surface of a metal is called the work function ϕ . ~~and~~ ^{It} depends largely on the nature of the metal. Thermionic emission is analogous to boiling off molecules from the surface of a liquid and the workfunction can be compared with the specific latent heat of vaporization.

Hot cathodes

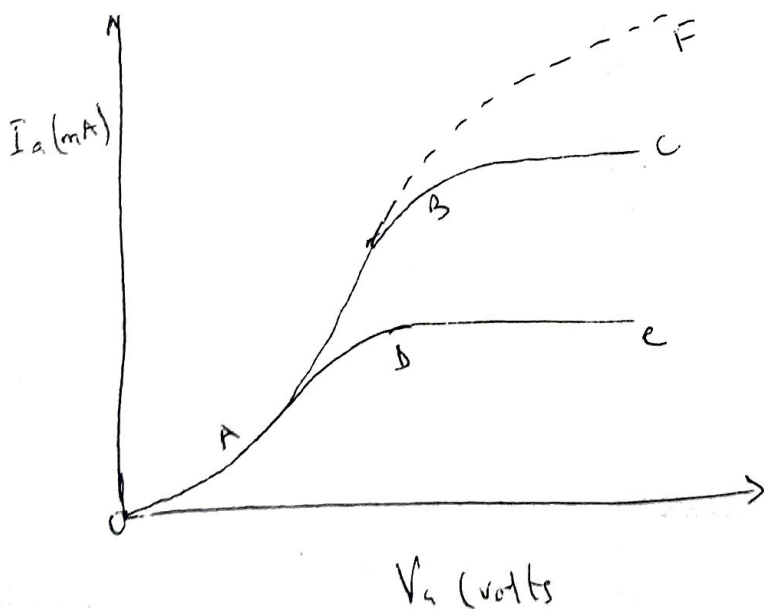
In many electronic devices, the cathode is heated and acts as the electron-emitter. Such devices are said to have thermionic or 'hot' cathode. The smaller the workfunction of a metal, the lower the temperature at which it releases electrons, any metal ~~that~~ ^{that} gives thermionic emission, in most cases, the temperature is to be ~~near~~ too near the melting point. Two emitting metals we use for hot cathodes are

(1) Tungsten This metal has a workfunction of 4.5 eV and operates at the rather high temperature of 2200°C, i.e. white heat, its melting point is about 3300°C. A large emission is possible and it is used chiefly in high-power radio transmitting valves.

anode voltage. The H.T can be a.c or d.c and varies from 114 Volts for a battery type valve to 6.3 Volts more for a mains-type valve. The H.T must be d.c. and variable up to 100 or 200 Volts according to the valve.

The anode current I_a is measured by a milliammeter and the anode voltage V_a by a voltmeter. The filament is connected to H.T negative to provide a complete circuit for the anode current.

To obtain the characteristic curve, V_a is increased by steps and the corresponding values of I_a recorded. If the results are plotted, a curve is obtained similar to OABC as in the diagram below.

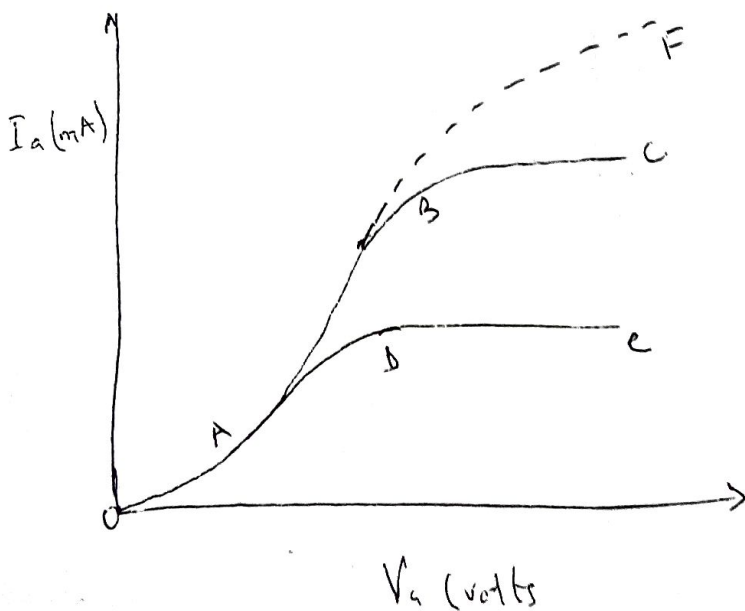


characteristic curve graph

It can be explained as follows. When V_a is zero, the electrons emitted by the filament tends to ~~cluster~~ cluster round it since their emission velocities are small.

anode voltage. The H.T can be a.c. or d.c. and varies from 1.4 Volts for a battery type valve to 6.3 Volts more for a mains-type valve. The H.T must be d.c. and variable up to 100 or 200 Volts according to the valve. The anode current I_a is measured by a milliammeter and the anode voltage V_a by a voltmeter. The filament is connected to H.T negative to provide a complete circuit for the anode current.

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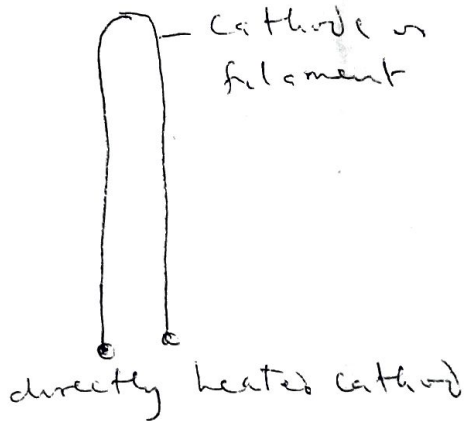
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coated valves saturate less abruptly due to the emission depending on the electric field at the cathode surface as well as ~~the~~ on the cathode temperature. They follow a curve similar to CF in the characteristic curve graph

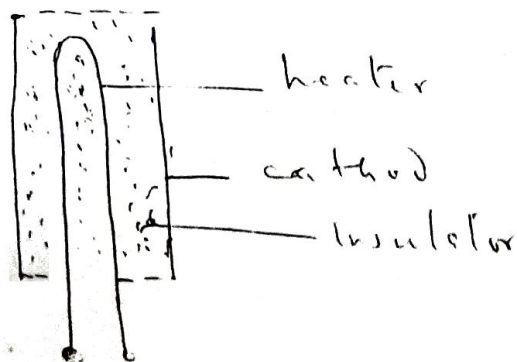
OXIDE-COATED METALS. A mixture of the oxides of barium and strontium has a work function of approximately 1 eV and give copious emission at 750°C i.e. dull red heat. Most thermionic devices employ this type of emitter.

Its cathodes are heated electrically either directly or indirectly.

In the direct heating, a current is passed through the cathode itself, which is in the form of a wire or filament made of tungsten or tungsten coated with oxide mixture.



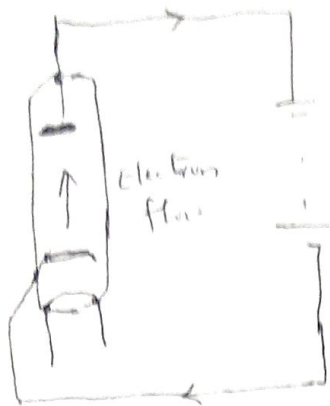
Indirect heating is possible with oxide-coated cathodes.



Indirect heated cathode.

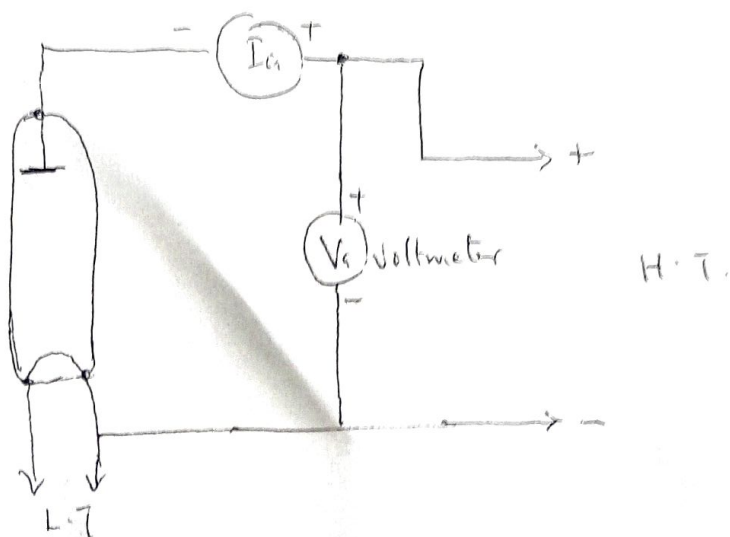
Here the cathode consists of a thin hollow nickel tube on which the oxide is sprayed. The cathode is heated by passing a current through the fine tungsten wire, called the heater, inside the tube.

Characteristics of a diode valve



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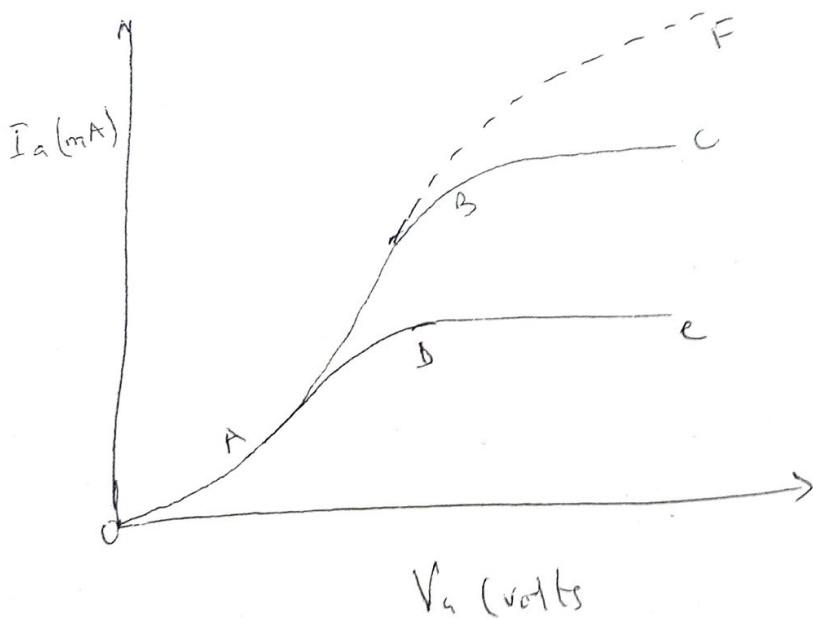
characteristic curves show how the anode current depends on the anode voltage i.e. the p.d. between the anode and cathode and may be determined experimentally for a directly heated diode using the circuit diagram below



As with most valve circuits, two voltages are required - a low tension or L.T. supply to maintain heating current through the filament and a high tension or H.T. supply to provide

anode voltage. The HT can be ac or dc and varies from 1.4 Volts for a battery type valve to 6.3 Volts more for a mains-type valve. The HT must be dc and variable up to 100 or 200 Volts according to the valve. The anode current I_a is measured by a milliammeter and the anode voltage V_a by a voltmeter. The filament is connected to H.T. negative to provide a complete circuit for the anode current.

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